N-Azolylmethyl Ketones as Building Blocks in Heterocyclic Synthesis: Synthesis of New Polyfunctionally Substituted Azolylarylazophenols, Azolylpyridones and Azolylthiophenes

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The title compounds 1a-b and 2 reacted with 2-arylhydrazonopropanals 3a-c to yield polyfunctionally substituted azolylarylazophenols 5 and 8. The reaction of 1b and 2 with phenylisothiocyanate in the presence of α -haloketones afforded the azolylthiophenes 12a,b and 13a,b. The reaction of 20 with α -haloketone afforded 5-benzotriazol-1-yl-6-methyl-2-(2-oxopropylsulfanyl)nicotinonitrile 21 that was utilized as building blocks for the synthesis of condensed pyridines. Compound 21 was condensed with dimethylformamide dimethylacetal to yield thieno[2,3-b]pyridin-3-yl-N,N-dimethylformamidine derivative 22. This was further cyclized with sodium hydride to 1*H*-thieno[2,3-*b*; 4,5-*b*`]dipyridin-4-one derivative 23.

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The reactivity of N-functionally substituted alkylbenzotriazoles toward electrophiles has been extensively utilized in the last decade by Katritzky et al. [1,2] and Elnagdi et al. [3-7] in the syntheses of polyfunctionally substituted heterocycles. Recently we have shown that N-functionally substituted pyrazolyl [8], N-1,2,4-triazolyl and N-benzimidazolyl-methyl ketones [9], also furnished carbanions under mild conditions. This enabled ready coupling with aromatic diazonium salts and condensation with dimethylformamide dimethylacetal. In conjunction with this work we report our finding on the reactivity of azolylmethyl ketones of types 1 and 2 (Scheme 1) toward carbon electrophiles in neutral and basic conditions.

Thus, it has been found that **1a,b** and **2** react with 2-arylhydrazonopropanals 3a-c to yield products of condensation via elimination of two water molecules. These can thus be formulated as 5 or its positional isomer 7 formed most likely *via* the presumed intermediates **4** and **6** respectively (Scheme 2). Structure 5 has been established based on the ¹H NMR spectra, as it revealed two aryl singlets at $\delta \sim 7.76$ and 8.34 ppm. This excludes completely structure 7 that





could have resulted, as it should display two doublets in its ¹H NMR spectrum. Similarly, compound 2 reacted with 3ac to yield **8a-c**. Similar to the recently reported behavior of 1a toward phenylisothiocyanate [7], compound 1b reacted with phenylisothiocyanate and α -haloketones in dimethyl formamide in the presence of potassium hydroxide to yield the azolylthiophenes 11a,b (Scheme 3). These are believed to be formed *via* intermediacy of 9 and 10a,b. The possible formation of thiazoles 12a,b was readily excluded based on ¹H NMR which indicated in each case the presence of NH signal at $\delta \sim 9.5$ ppm. Similarly, compound 2 reacted with phenylisothiocyanate and α -haloketones under similar reaction conditions to yield 13a,b. It is of value to report that in earlier work [7], thiazoles were formed from reaction of 1c with the same reagents under the same reaction conditions, demonstrating dependency of the final product on the nature of the utilizing ketone.

Scheme 3



In light of these results, another electrophile was treated under an alkaline condition (Scheme 4). In fact, the reaction of **1c** with benzylidenemalononitrile in ethanolic piperidine has afforded a product of molecular formula $C_{24}H_{15}N_5O$. This is assumed to be **17**, formed *via* rearrangement of the initially formed 4*H*-pyrane **15** into the 1,2,3,4-tetrahydropyridine **16** that auto-oxidized under reaction conditions into the pyridone **17**. Ready oxidation of dihydroazines into azines has been reported earlier under mild conditions [10].

To increase the synthetic scope on the utility of these types of substrates, chemical modification of the synthon



1a was then investigated to access complex heterocyclic systems (Schemes 5 and 6). Accordingly, dimethylformamide dimethylacetal condensed with 1a yielding the enaminone 18 [7]. Compound 18 reacted with cyanothioacetamide to yield products that were formulated as 20 rather than 19. ¹H NMR revealed pyridine H₄ proton at $\delta = 8.01$ ppm as singlet. For compound **19**, this proton should appear as a doublet confirming this structural assignment. Compound 20 reacted with chloroacetone to yield 5,6-disubstituted-2-(2-oxopropy-Isulfanyl)-nicotinonitrile 21, this is in contrast to the reported direct formation of 1-(3-amino-5-benzotriazol-1-yl-6methylthieno[2,3-b]pyridin-2-yl) derivative on reacting 20 with phenacyl bromide [11]. Compound 21 condensed with dimethylformamide dimethylacetal affording the thieno[2,3b]pyridin-3-yl-N,N-dimethylformamidine derivative 22. This was further cyclized with sodium hydride to thieno-[2,3-b:4,5-b']dipyridin-4-one derivative 23. It is of value to report that H₀ in 23 appeared at $\delta = 9.04$ ppm as a singlet, this low field shift is a result of deshielding by the anisotropic effect of the N-lone-pair.



One may thus conclude that attaching a methylene ketone to the ring nitrogen in benzotriazole, benzimidazole, triazole and imidazole activate the methylene moiety toward carbon electrophiles. Thus enabling the



synthesis of a variety of polyfunctionally substituted azolyl and azinyl heteroaromatics. This is in contrast with the reported ready elimination of the benzotriazole moiety during reaction of alkyl benzotriazoles with electrophilic reagents, no such elimination has been observed in our systems.

EXPERIMENTAL

All melting points are uncorrected. IR spectra were recorded in KBr with a Pye Unicam SP 1100 spectrophotometer. ¹H NMR spectra were recorded on a Varian EM-390 400 MHz spectrometer in deuteriochloroform or [²H₆] DMSO as solvent and TMS as internal standard; chemical shifts are reported in δ units (ppm). Mass spectra were measured on MS 30 and MS 9 (AEI) instruments operating at 70 eV. Microanalyses were performed on a LECO CHNS-932. Compounds **1a,b** and **2** were prepared following published procedure [12,13,9].

2-Benzotriazol-1-yl-1-phenylethanone (1c).

A mixture of benzotriazole (1.19 g, 10 mmol), and phenacyl bromide (1.99 g, 10 mmol), in DMF (20 ml) and in the presence of sodium hydroxide (0.05 g) was heated under reflux for 2 h. The solvent was reduced under vacuum, poured into water and neutralized with diluted hydrochloric acid solution to deposit a solid, which was crystallized from ethanol. Yield: 1.70 g (72 %); mp 115°. ir (KBr) $\nu_{max} = 1689$ (CO) cm⁻¹; MS (EI, 70 EV): m/z (%) = 237 [M⁺]; ¹H nmr (deuteriochloroform): $\delta_{\rm H} = 6.11$ (s, 2H, CH₂), 7.37-7.41 (m, 2H, arom. H), 7.47-7.57 (m, 3H, arom. H), 7.66-7.70 (m, 1H, arom. H), 8.06-8.11 (m, 3H, arom. H).

Anal. Calcd. for C₁₄H₁₁N₃O (237.25): C, 70.87; H, 4.67; N, 17.71. Found C, 70.90; H, 4.53; N, 17.69.

General Procedure for the Preparation of Compounds 5a-f.

To a stirred suspension of each of compound **3a-c** (10 mmol) in ethanol (20 ml) containing potassium hydroxide (0.56 g, 10 mmol), each of compound **1a,b** (10 mmol) was added. The reaction mixture was heated under reflux for 2 h. The solvent was evaporated under reduced pressure and the remaining product was triturated with water and neutralized with hydrochloric acid to deposit a solid, which was crystallized from ethanol/dioxane (3:1).

4-Benzotriazol-1-yl-6-phenylazobiphenyl-3-ol (5a).

This compound was obtained in 77% yield (3.01 g); mp. 217°; ir (KBr) $\nu_{max} = 3432$ (OH) cm⁻¹; MS (EI, 70 EV): *m/z* (%) = 391 [M⁺]; ¹H nmr (deuteriochloroform): $\delta_{\rm H} = 7.37$ (s, 1H, H₂), 7.48-7.56 (m, 7H, arom. H), 7.59-7.64 (m, 4H, arom. H), 7.67-7.69 (m, 2H, arom. H), 7.99 (s, 1H, H₅), 8.19 (d, 1H, *J* = 8.4 Hz, benzotriazolyl-H), 11.55 (s, 1H, OH); ¹³C NMR (deuteriochloroform): $\delta_{\rm C} = 156.01$, 152.99, 145.16, 144.07, 143.90, 142.46, 138.44, 135.00, 131.76, 131.37, 130.24, 128.79, 124.20, 124.04, 123.39, 122.97, 120.51, 118.92, 115.64, 111.85.

Anal. Calcd. for $C_{24}H_{17}N_5O$ (391.42): C, 73.64; H, 4.38; N, 17.89. Found C, 73.64; H, 4.53; N, 17.62.

2-Benzotriazol-1-yl-4-phenylazo-5-thiophen-2-yl-phenol (5b).

This compound was obtained in 92% yield (3.65 g); mp. 280°; ir (KBr) $v_{max} = 3422$ (OH) cm⁻¹; MS (EI, 70 EV): m/z (%) = 397 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 7.28$ (dd, 1H, J = 4 Hz, thienyl H₄), 7.47-7.51 (m, 1H, thienyl H₃), 7.56 (d, 1H, J = 8.0 Hz, benzotriazolyl-H), 7.59-7.65 (m, 5H, Ph), 7.68 (s, 1H, H₂), 7.86 (dd, 1H, J = 4.6 Hz, thienyl H₅), 7.93 (m, 2H, benzotriazolyl-H), 7.97 (s, 1H, H₅), 8.18 (d, 1H, J = 8.2 Hz, benzotriazolyl H), 11.68 (s, 1H, OH); ¹³C nmr (DMSO): $\delta_{\rm C} = 155.68$, 153.01, 145.91, 141.17, 138.19, 137.15, 134.37, 132.28, 132.17, 130.42, 129.22, 128.98, 128.35, 125.17, 124.43, 124.11, 120.22, 116.62, 116.32, 112.63.

Anal. Calcd. for C₂₂H₁₅N₅OS (397.38): C, 66.49; H, 3.80; N, 17.63; S, 8.05. Found C, 66.72; H, 4.08; N, 17.52; S, 7.84.

2-Benzotriazol-1-yl-5-furan-2-yl-4-phenylazo-phenol (5c).

This compound was obtained in 91% yield (3.46 g); mp. 260°; ir (KBr) $v_{max} = 3435$ (OH) cm⁻¹; MS (EI, 70 EV): *m/z* (%) = 381 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 6.76-6.78$ (m, 1H, furyl H₄), 7.08 (d, 1H, *J* = 3.4 Hz, furyl H₃), 7.47-7.51 (m, 1H, furyl H₅), 7.56-7.64 (m, 5H, Ph), 7.76 (s, 1H, H₆), 7.89 (m, 2H, benzotriazolyl H), 7.93 (s, 1H, H₃), 8.01-8.02 (m, 1H, benzotriazolyl-H), 8.17 (d, 1H, *J* = 8.1 Hz, benzotriazolyl H), 11.54 (s, 1H, OH); ¹³C nmr (DMSO): $\delta_{\rm C} = 155.56$, 153.08, 149.58, 145.89, 145.22, 140.97, 134.35, 132.57, 132.18, 130.49, 128.95, 125.16, 123.94, 123.71, 120.21, 116.44, 116.38, 114.02, 113.97, 112.65.

Anal. Calcd. for C₂₂H₁₅N₅O₂ (381.38): C, 69.28; H, 3.96; N, 18.36. Found C, 69.18; H, 4.11; N, 18.13.

4-Benzimidazol-1-yl-6-phenylazo-biphenyl-3-ol (5d).

This compound was obtained in 80% yield (3.12 g); mp. 288°; ir (KBr) $v_{max} = 3420 \text{ cm}^{-1}$ (OH); MS (EI, 70 EV): m/z (%) = 390 [M⁺]; ¹H nmr (DMSO): $\delta_{\text{H}} = 7.27$ (s, 1H, H₂), 7.29-7.30 (m, 2H, arom. H), 7.38-7.40 (m, 1H, arom. H), 7.46-7.57 (m, 8H, arom. H), 7.64-7.66 (m, 2H, arom. H), 7.77-7.79 (m, 1H, arom. H), 7.90 (s, 1H, H₅), 8.47 (s, 1H, benzimidazolyl H₂), 11.20 (s, 1H, OH). *Anal.* Calcd. for C₂₅H₁₈N₄O (390.43): C, 76.90; H, 4.65; N, 14.35. Found C, 77.09; H, 4.73; N, 14.26.

2- Benzimidazol-1-yl-4-phenylazo-5-thiophen-2-yl-phenol (5e).

This compound was obtained in 85% yield (3.36 g); mp. 276°; ir (KBr) $\nu_{max} = 3439$ (OH) cm⁻¹; MS (EI, 70 EV): m/z (%) = 396 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 7.25-7.31$ (m, 3H, arom. H), 7.37-7.40 (m, 1H, thienyl H₄), 7.54 (d, 1H, J = 4.8 Hz, thienyl H₃), 7.62-7.35 (m, 4H, arom. H), 7.76-7.78 (m, 1H, arom. H), 7.84 (d, 1H, J = 4.6 Hz, thienyl H₅), 7.88 (s, 1H, H-3), 7.92 (d, 2H, J = 8.0 Hz, benzimidazolyl H), 8.46 (s, 1H, benzimidazolyl H₂), 11.40 (s, 1H, OH); ¹³C nmr (DMSO): $\delta_{\rm C} = 156.46$, 153.03, 145.10, 143.82, 141.11,

138.40, 136.16, 134.91, 132.02, 130.40, 128.89, 128.28, 124.26, 124.09, 123.03, 121.96, 120.49, 118.48, 116.49, 115.76, 111.87.

Anal. Calcd. for C₂₃H₁₆N₄OS (396.38): C, 69.69; H, 4.07; N, 14.14; S, 8.07. Found C, 69.63; H, 4.18; N, 13.95; S, 7.83.

2- Benzimidazol-1-yl-5-furan-2-yl-4-phenylazo-phenol (5f).

This compound was obtained in 68% yield (2.58 g); mp. 273°; ir (KBr) $\nu_{max} = 3435$ (OH) cm⁻¹; MS (EI, 70 EV): *m/z* (%) = 380 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 6.75$ (t, 1H, *J* = 3.4 Hz, furyl H₄), 7.03 (d, 1H, *J* = 3.4 Hz, furyl H₃), 7.28-7.30 (m, 2H, arom. H), 7.36-7.38 (m, 1H, furyl H₅), 7.55-7.63 (m, 3H, arom. H), 7.71 (s, 1H, H-6), 7.76-7.82 (m, 1H, benzoimidazolyl H), 7.84-7.87 (m, 3H, benzoimidazolyl H), 7.98 (s, 1H, H-3), 8.46 (s, 1H, benzoimidazolyl H₂), 11.55 (s, 1H, OH).

Anal. Calcd. for C₂₃H₁₆N4O₂ (380.39): C, 72.62; H, 4.24; N, 14.73. Found C, 72.88; H, 4.25; N, 14.56.

General Procedure for the Preparation of Compounds 8a-c.

Similar reaction conditions as for compounds **7a-f** using compound **2** (10 mmol) were utilized. The target product **8a-c** was crystallized from ethanol/dioxane (3:1).

6-Phenylazo-4-[1,2,4]-triazol-1-yl-biphenyl-3-ol (8a).

This compound was obtained in 88% yield (3.0 g); mp. 230°; ir (KBr) $\nu_{max} = 3415$ (OH) cm⁻¹; MS (EI, 70 EV): m/z (%) = 341 [M⁺]; ¹H nmr (deuteriochloroform): $\delta_{H} = 7.38$ (s, 1H, H₂), 7.46-7.51 (m, 6H, arom. H), 7.55-7.57 (m, 2H, arom. H), 7.79-7.81 (m, 2H, arom. H), 8.11 (s, 1H, H₅), 8.27 (s, 1H, triazolyl H₄), 8.89 (s, 1H, triazolyl H₂), 10.61 (s, 1H, OH).

Anal. Calcd. for C₂₀H₁₅N₅O (341.36): C, 70.37; H, 4.43; N, 20.52. Found C, 70.31; H, 4.58; N, 9.87.

4-Phenylazo-5-thiophen-2-yl-2-[1,2,4]-triazol-1-yl-phenol (8b).

This compound was obtained in 85% yield (3.0 g); mp. 271°; ir (KBr) $\nu_{max} = 3435$ (OH) cm⁻¹; MS (EI, 70 EV): m/z (%) = 396 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 7.23$ (t, 1H, J = 4 Hz, thienyl H₄), 7.54-7.64 (m, 5H, arom. H), 7.82 (d, 1H, J = 4.2 Hz, thienyl H₅), 7.94-7.96 (m, 2H, arom. H), 8.15 (s, 1H, H₃), 8.32 (s, 1H, triazolyl H₄), 9.15 (s, 1H, triazolyl H₂), 11.71 (s, 1H, OH).

Anal. Calcd. for C₁₈H₁₃N₅OS (347.32): C, 62.24; H, 3.78; N, 20.17; S, 9.21. Found C, 62.48; H, 3.99; N, 19.95; S, 9.11.

5-Furan-2-yl- 4-phenylazo-2-[1,2,4]-triazol-1-yl-phenol (8d).

This compound was obtained in 87% yield (2.85 g); mp. 236°; ir (KBr) $\nu_{max} = 3435$ (OH) cm⁻¹; MS (EI, 70 EV): m/z (%) = 380 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 6.79$ (t, 1H, J = 3.6 Hz, furyl H₄), 7.00 (d, 1H, J = 3.6 Hz, furyl H₃), 7.54-7.65 (m, 4H, arom. H), 7.81-7.96 (m, 3H, arom. H), 8.12 (s, 1H, H₃), 8.26 (s, 1H, triazolyl H-4), 9.16 (s, 1H, triazolyl H₂), 11.70 (s, 1H, OH).

Anal. Calcd. for C₁₈H₁₃N₅O₂ (331.32): C, 65.25; H, 3.96; N, 21.14. Found C, 65.09; H, 4.07; N, 21.09.

General Procedure for the Preparation of Compounds **11a,b** and **13a,b**.

A stirred suspension of each of compound **1b** and 2 (10 mmol) and phenyl isothiocyanate (1.35 g, 10 mmol) in DMF (20 ml) in the presence of potassium hydroxide (0.06 g); α -haloketone (10 mmol) is added to the stirred solution after 10 h and the reaction mixture is heated under reflux for 4 h. The solvent was removed and the residue was triturated with water and neutralized with hydrochloric acid. The solid product was crystallized from ethanol. 1-(4-Benzimidazol-1-yl-3-methyl-5-phenylaminothiophen-2-yl)ethanone **11a**.

This compound was obtained in 75% yield (2.60 g); mp. 240°; ir (KBr) $\nu_{max} = 3086$ (NH), 1614 (CO) cm⁻¹; MS (EI, 70 EV): m/z (%) = 347 [M⁺]; ¹H nmr (deuteriochloroform): $\delta_{\rm H} = 2.10$ (s, 3H, CH₃), 2.52 (s, 3H, COCH₃), 7.10-7.14 (m, 1H, arom. H), 7.17-7.23 (m, 2H, arom. H), 7.36-7.41 (m, 3H, arom. H), 7.46-7.51 (m, 3H, arom. H), 8.96 (s, 1H, benzimidazolyl-H), 9.57 (br s, 1H, NH); ¹³C nmr (deutereochloroform): $\delta_{\rm C} = 189.9$ (CO), 151.8, 143.6, 143.0, 142.4, 141.3, 134.0, 130.0, 124.5, 124.1, 123.7, 120.2, 119.8, 119.2, 115.7, 110.7, 29.5, 14.5.

Anal. Calcd. for C₂₀H₁₇N₃OS (347.36): C, 69.15; H, 4.93; N, 12.10; S, 9.21. Found C, 68.95; H, 5.01; N, 12.08; S, 9.00.

(4-Benzimidazol-1-yl-3-methyl-5-phenylaminothiophen-2-yl)-phenylmethanone (**11b**).

This compound was obtained in 69% yield (2.82 g); mp. 150°; ir (KBr) $\nu_{max} = 3055$ (NH), 1606 (CO) cm⁻¹; MS (EI, 70 EV): m/z (%) = 409 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 1.83$ (s, 3H, CH₃), 7.01-7.03 (m, 1H, arom. H), 7.26-7.32 (m, 7H, arom. H), 7.50-7.57 (m, 3H, arom. H), 7.69-7.71 (m, 2H, arom. H), 7.76-7.78 (m, 1H, arom. H), 8.38 (s, 1H, benzimidazolyl H₂), 9.36 (br s, 1H, NH); ¹³C nmr (DMSO): $\delta_{\rm C} = 188.3$ (CO), 153.3, 145.6, 144.6, 142.5, 141.8, 140.8, 134.6, 132.5, 130.31, 129.4, 128.8, 124.7, 124.3, 123.6, 120.1, 119.8, 118.4, 117.4, 111.5, 15.0.

Anal. Calcd. for C₂₅H₁₉N₃OS (409.43): C, 73.33; H, 4.68; N, 10.26; S, 7.81. Found C, 73.11; H, 4.74; N, 10.39; S, 7.60.

1-(3-Methyl-5-phenylamino-4-[1,2,4]-triazol-1-yl-thiophen-2-yl)ethanone (**13a**).

This compound was obtained in 80% yield (2.40 g); mp. 185°; ir (KBr) $v_{max} = 3064$ (NH), 1613 (CO) cm⁻¹; MS (EI, 70 EV): m/z (%) = 298 [M⁺]; ¹H nmr (deuteriochloroform): $\delta_{\rm H} = 2.05$ (s, 3H, CH₃), 2.43 (s, 3H, COCH₃), 7.04-7.07 (m, 1H, arom. H), 7.27-7.36 (m, 5H, arom. H), 8.26 (s, 1H, triazolyl H₄), 8.77 (s, 1H, triazolyl H₂), 9.16 (br s, 1H, NH).

Anal. Calcd. for C₁₅H₁₄N₄OS (298.29): C, 60.39; H, 4.73; N, 18.78; S, 10.73. Found C, 60.59; H, 4.79; N, 18.80; S, 10.71.

(3-Methyl-5-phenylamino-4-[1,2,4]-triazol-1-yl- -thiophen-2-yl)phenyl-methanone (**13b**).

This compound was obtained in 72% yield (2.60 g); mp. 190°; ir (KBr) $v_{max} = 3069$ (NH), 1617 (CO) cm⁻¹; MS (EI, 70 EV): m/z (%) = 360.36 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 1.88$ (s, 3H, CH₃), 7.05-7.07 (m, 1H, arom. H), 7.26-7.35 (m, 7H, arom. H), 7.49-7.53 (m, 3H, arom. H), 7.57-7.59 (m, 1H, arom. H), 7.65-7.67 (m, 2H, arom. H), 8.26 (s, 1H, triazolyl H₄), 8.80 (s, 1H, triazolyl H₂), 9.29 (br s, 1H, NH); ¹³C nmr (DMSO): $\delta_{\rm C} = 188.47$, 153.33, 153.28, 147.62, 144.20, 141.51, 140.71, 132.54, 130.45, 129.44, 128.75, 124.66, 120.12, 119.31, 117.99, 14.87 (CH₃).

Anal. Calcd. for C₂₀H₁₆N₄OS (360.36): C, 66.66; H, 4.48; N, 15.55; S, 8.88. Found C, 67.00; H, 4.48; N, 15.40; S, 9.06.

5-Benzotriazol-1-yl-2-oxo-4,6-diphenyl-1,2-dihydropyridine-3-carbonitrile (**17**).

Benzylidenemalononitrile (1.54 g, 10 mmol) was added to a stirred suspension of compound 1c (2.37 g, 10 mmol) in ethanol (20 ml) in the presence of sodium hydroxide (0.5 g). The reaction mixture was heated under reflux for 2 h. The solvent was reduced under vacuum; poured onto water and the residue was neutralized

with diluted hydrochloric acid solution to deposit a solid, which was crystallized from dioxane. Yield: 2.40 g (62 %); mp. > 300°; ir (KBr) $v_{max} = 3060$ (NH), 2229 (CN), 1672 (CO) cm⁻¹; MS (EI, 70 EV): m/z (%) = 389 [M⁺]; ¹H nmr (DMSO): $\delta_{\rm H} = 7.00$ -7.26 (m, 10H, arom. H), 7.33-7.40 (m, 2H, benzotriazolyl-H), 7.60 (d, 1H, J = 8.6 Hz, benzotriazolyl-H), 7.78 (d, 1H, J = 8.6 Hz, benzotriazolyl-H), 13.46 (br s, 1H, NH).

Anal. Calcd. for $C_{24}H_{15}N_5O$ (389.40): C, 74.02; H, 3.88; N, 17.99. Found C, 74.27; H, 4.02; N, 18.21.

5-Benzotriazol-1-yl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile (**20**).

Cyanothioacetamide (1 g, 10 mmol) was added to a stirred suspension of the enaminone **18** (2.30 g, 10 mmol) in ethanol (20 ml) in the presence of few drops of triethyl amine. The reaction mixture was heated under reflux for 2 h. The solvent was reduced under vacuum; poured onto water and the residue was neutralized with diluted hydrochloric acid solution to deposit a solid, which was crystallized from ethanol. Yield: 1.70 g (64 %); mp. 290°; ir (KBr) $v_{max} = 3013$ (NH), 2224 (CN) cm⁻¹; MS (EI, 70 EV): m/z (%) = 267 [M⁺]; ¹H nmr (deuteriochloroform): $\delta = 2.38$ (s, 3H, CH₃), 7.36-7.38 (m, 1H, benzotriazolyl-H), 7.52-7.59 (m, 1H, benzotriazolyl-H), 8.01 (s, 1H, pyridyl H₄), 8.18-8.27 (m, 1H, benzotriazolyl-H), 13.65 (br s, 1H, NH).

Anal. Calcd. for C₁₉H₁₄N₂S (267.24): C, 58.42; H, 3.39; N, 26.21; S, 11.97. Found: C, 58.51; H, 3.54; N, 25.95; S, 11.82.

5-Benzotriazol-1-yl-6-methyl-2-(2-oxopropylsulfanyl)nicotinonitrile (**21**).

Chloroacetone (0.92 g, 10 mmol) was added to a stirred suspension of compound 20 (2.67 g, 10 mmol) in DMF (20 ml) in the presence of potassium hydroxide (0.5 g). The reaction mixture was heated under reflux for 6 h. The solvent was reduced under vacuum; poured onto water and the residue was neutralized with diluted hydrochloric acid solution to deposit a solid, which was crystallized from ethanol/dioxane (1:2). Yield: 2.60 g (80 %); mp. 259°; ir (KBr) $v_{\text{max}} = 2228$ (CN), 1676 (CO) cm⁻¹; MS (EI, 70 EV): m/z (%) = 323 [M⁺]; ¹H nmr (deuteriochloroform): $\delta_{\rm H} = 2.46$ (s, 1H, CH₃), 2.48 (s, 1H, COCH₃), 6.79 (s, 2H, CH₂), 7.34 (d, 1H, J = 8.4 Hz, benzotriazolyl-H), 7.50 (t, 1H, J = 8.6Hz, benzotriazolyl-H), 7.59 (t, 1H, J = 8.6 Hz, benzotriazolyl-H), 8.08 (s, 1H, pyridyl H₄), 8.20 (d, 1H, J = 8.4 Hz, benzotriazolyl-H); ¹³C nmr (DMSO): $\delta_{\rm C} = 192.37$ (CO), 160.68, 157.78, 148.15, 145.86, 134.42, 131.60, 129.82, 128.79, 125.73, 120.58, 111.62, 106.04, 31.57 (CH₂), 29.89 (CH₃), 22.11 (COCH₃).

Anal. Calcd. for C₁₆H₁₃N₅OS (323.30): C, 59.44; H, 4.05; N, 21.66; S, 9.90. Found C, 59.64; H, 4.07; N, 21.66; S, 9.68.

N'-(2-Acetyl-5-benzotriazol-1-yl-6-methyl-thieno[2,3-*b*]pyridin-3-yl)-*N*,*N*-dimethylformamidine (**22**).

To a suspension of compound **20** (3.23 g, 10 mmol) in xylene (20 ml), dimethylformamide dimethylacetal (1,19 g, 10 mmol) was added. The reaction mixture was heated under reflux for 4 h. The solvent was removed and the residue cooled to deposit a solid, which was crystallized from dioxane/ethanol (1:1). Yield: 2.50 g (65 %); mp. 264°; ir (KBr) $v_{max} = 1636$ (CO) cm⁻¹; MS (EI, 70 EV): *m/z* (%) = 378 [M⁺]; ¹H nmr (DMSO): $\delta_H = 2.29$ (s, 3H, CH₃), 2.55 (s,

3H, COCH₃), 3.01 (s, 6H, NMe₂), 7.53 (t, 1H, J = 8.6 Hz, benzotriazolyl-H), 7.61-7.67 (m, 2H, benzotriazolyl-H), 7.97 (s, 1H, H₄), 8.23 (d, 1H, J = 8.6 Hz, benzotriazolyl-H), 8.42 (s, 1H, amidine-H). *Anal.* Calcd. for C₁₉H₁₈N₆OS (378.38): C, 60.31; H, 4.80; N,

22.21; S, 8.46. Found C, 60.14; H, 4.88; N, 21.98; S, 8.21.

8-Benzotriazol-1-yl-7-methyl-1*H*-thieno[2,3-*b*; 4,5-*b*']dipyridin-4-one (**23**).

To a suspension of compound **22** (3.78 g, 10 mmol) in dioxane (20 ml), sodium hydride (0.4 g, 60 % dispersion) was added. The reaction mixture was heated under reflux for 6 h. The solvent was reduced under vacuum; poured into water and the residue was neutralized with diluted hydrochloric acid solution to deposit a solid, which was crystallized from dioxane/ethanol (1:3). Yield: 2.0 g (60 %); mp. > 300 °; ir (KBr) $v_{max} = 33.92$ (OH) cm⁻¹; MS (EI, 70 EV): m/z (%) = 333 [M⁺]; ¹H nmr (DMSO): $\delta_{H} = 2.48$ (s, 3H, CH₃), 3.84 (br s, 1H, NH), 6.62 (d, 1H, J = 8.4 Hz, H₃), 7.57 (t, 1H, J = 8.6 Hz, benzotriazolyl-H), 7.68 (t, 1H, J = 8.6 Hz, benzotriazolyl-H), 7.76 (d, 1H, J = 8.4 Hz, H₂), 8.26-8.30 (m, 2H, benzotriazolyl-H), 9.04 (s, 1H, H₉); ¹³C nmr (DMSO): $\delta_{C} = 161.23$ (CO), 156.97, 145.92, 142.65, 134.41, 130.62, 129.93, 129.50, 128. 45, 127.65, 126.21, 125.82, 124.79, 120.65, 111.97, 111.59, 22.23.

Anal. Calcd. for C₁₇H₁₁N5OS (333.38): C, 61.26; H, 3.33; N, 21.01; S, 9.60. Found C, 61.14; H, 3.48; N, 20.92; S, 9.87.

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