# N -Azolylmethyl Ketones as Building Blocks in Heterocyclic <br> Synthesis: Synthesis of New Polyfunctionally Substituted <br> Azolylarylazophenols, Azolylpyridones and Azolylthiophenes 

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#### Abstract

The title compounds 1a-b and 2 reacted with 2 -arylhydrazonopropanals $3 \mathrm{a}-\mathrm{c}$ to yield polyfunctionally substituted azolylarylazophenols 5 and 8 . The reaction of 1 b and 2 with phenylisothiocyanate in the presence of $\alpha$-haloketones afforded the azolylthiophenes $12 \mathrm{a}, \mathrm{b}$ and $13 \mathrm{a}, \mathrm{b}$. The reaction of 20 with $\alpha$-haloketone afforded 5-benzotriazol-1-yl-6-methyl-2-(2-oxopropylsulfanyl)nicotinonitrile 21 that was utilized as building blocks for the synthesis of condensed pyridines. Compound 21 was condensed with dimethylformamide dimethylacetal to yield thieno[2,3-b]pyridin-3-yl- $N, N$-dimethylformamidine derivative 22 . This was further cyclized with sodium hydride to $1 H$-thieno[2,3-b; 4,5-b ] ]dipyridin-4-one derivative 23.


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The reactivity of $N$-functionally substituted alkylbenzotriazoles toward electrophiles has been extensively utilized in the last decade by Katritzky et al. [1,2] and Elnagdi et al. [3-7] in the syntheses of polyfunctionally substituted heterocycles. Recently we have shown that $N$-functionally substituted pyrazolyl [8], $N$-1,2,4-triazolyl and $N$-benzimi-dazolyl-methyl ketones [9], also furnished carbanions under mild conditions. This enabled ready coupling with aromatic diazonium salts and condensation with dimethylformamide dimethylacetal. In conjunction with this work we report our finding on the reactivity of azolylmethyl ketones of types 1 and 2 (Scheme 1) toward carbon electrophiles in neutral and basic conditions.
Thus, it has been found that $\mathbf{1 a , b}$ and $\mathbf{2}$ react with 2 -arylhydrazonopropanals 3a-c to yield products of condensation
via elimination of two water molecules. These can thus be formulated as $\mathbf{5}$ or its positional isomer $\mathbf{7}$ formed most likely via the presumed intermediates $\mathbf{4}$ and $\mathbf{6}$ respectively (Scheme 2). Structure 5 has been established based on the ${ }^{1} \mathrm{H}$ NMR spectra, as it revealed two aryl singlets at $\delta \sim 7.76$ and 8.34 ppm . This excludes completely structure 7 that


Scheme 2

could have resulted, as it should display two doublets in its ${ }^{1} \mathrm{H}$ NMR spectrum. Similarly, compound 2 reacted with 3ac to yield 8a-c. Similar to the recently reported behavior of 1a toward phenylisothiocyanate [7], compound $\mathbf{1 b}$ reacted with phenylisothiocyanate and $\alpha$-haloketones in dimethyl formamide in the presence of potassium hydroxide to yield the azolylthiophenes 11a,b (Scheme 3). These are believed to be formed via intermediacy of $\mathbf{9}$ and 10a,b. The possible formation of thiazoles 12a,b was readily excluded based on ${ }^{1} \mathrm{H}$ NMR which indicated in each case the presence of NH signal at $\delta \sim 9.5 \mathrm{ppm}$. Similarly, compound 2 reacted with phenylisothiocyanate and $\alpha$-haloketones under similar reaction conditions to yield $\mathbf{1 3 a}, \mathbf{b}$. It is of value to report that in earlier work [7], thiazoles were formed from reaction of $\mathbf{1 c}$ with the same reagents under the same reaction conditions, demonstrating dependency of the final product on the nature of the utilizing ketone.

Scheme 3



11a, $\mathrm{R}^{2}=\mathrm{CH}_{3}$
b, $\mathrm{R}^{2}=\mathrm{Ph}$


$\begin{aligned} \text { 13a, } \mathrm{R}^{2} & =\mathrm{CH}_{3} \\ \text { b, } \mathrm{R}^{2} & =\mathrm{Ph}\end{aligned}$

In light of these results, another electrophile was treated under an alkaline condition (Scheme 4). In fact, the reaction of 1c with benzylidenemalononitrile in ethanolic piperidine has afforded a product of molecular formula $\mathrm{C}_{24} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$. This is assumed to be $\mathbf{1 7}$, formed via rearrangement of the initially formed $4 H$-pyrane 15 into the 1,2,3,4-tetrahydropyridine $\mathbf{1 6}$ that auto-oxidized under reaction conditions into the pyridone 17. Ready oxidation of dihydroazines into azines has been reported earlier under mild conditions [10].

To increase the synthetic scope on the utility of these types of substrates, chemical modification of the synthon

Scheme 4


1a was then investigated to access complex heterocyclic systems (Schemes 5 and 6). Accordingly, dimethylformamide dimethylacetal condensed with 1a yielding the enaminone 18 [7]. Compound 18 reacted with cyanothioacetamide to yield products that were formulated as 20 rather than $19 .{ }^{1} \mathrm{H}$ NMR revealed pyridine $\mathrm{H}_{4}$ proton at $\delta=8.01 \mathrm{ppm}$ as singlet. For compound 19, this proton should appear as a doublet confirming this structural assignment. Compound 20 reacted with chloroacetone to yield 5,6-disubstituted-2-(2-oxopropy-lsulfanyl)-nicotinonitrile 21, this is in contrast to the reported direct formation of 1-(3-amino-5-benzotriazol-1-yl-6-methylthieno[2,3-b]pyridin-2-yl) derivative on reacting 20 with phenacyl bromide [11]. Compound 21 condensed with dimethylformamide dimethylacetal affording the thieno[2,3$b]$ pyridin-3-yl- $\mathrm{N}, \mathrm{N}$-dimethylformamidine derivative 22. This was further cyclized with sodium hydride to thieno-[2,3-b:4,5-b'] dipyridin-4-one derivative 23. It is of value to report that $\mathrm{H}_{9}$ in 23 appeared at $\delta=9.04 \mathrm{ppm}$ as a singlet, this low field shift is a result of deshielding by the anisotropic effect of the $N$-lone-pair.


One may thus conclude that attaching a methylene ketone to the ring nitrogen in benzotriazole, benzimidazole, triazole and imidazole activate the methylene moiety toward carbon electrophiles. Thus enabling the

Scheme 6

$\mathrm{Bt}=$ benzotriazol $-1-\mathrm{yl}$
synthesis of a variety of polyfunctionally substituted azolyl and azinyl heteroaromatics. This is in contrast with the reported ready elimination of the benzotriazole moiety during reaction of alkyl benzotriazoles with electrophilic reagents, no such elimination has been observed in our systems.

## EXPERIMENTAL

All melting points are uncorrected. IR spectra were recorded in KBr with a Pye Unicam SP 1100 spectrophotometer. ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Varian EM- 390400 MHz spectrometer in deuteriochloroform or $\left[{ }^{2} \mathrm{H}_{6}\right]$ DMSO as solvent and TMS as internal standard; chemical shifts are reported in $\delta$ units (ppm). Mass spectra were measured on MS 30 and MS 9 (AEI) instruments operating at 70 eV . Microanalyses were performed on a LECO CHNS-932. Compounds 1a,b and 2 were prepared following published procedure [12,13,9].

## 2-Benzotriazol-1-yl-1-phenylethanone (1c).

A mixture of benzotriazole ( $1.19 \mathrm{~g}, 10 \mathrm{mmol}$ ), and phenacyl bromide ( $1.99 \mathrm{~g}, 10 \mathrm{mmol})$, in DMF ( 20 ml ) and in the presence of sodium hydroxide ( 0.05 g ) was heated under reflux for 2 h . The solvent was reduced under vacuum, poured into water and neutralized with diluted hydrochloric acid solution to deposit a solid, which was crystallized from ethanol. Yield: 1.70 g (72 \%); $\mathrm{mp} 115^{\circ}$. ir (KBr) $v_{\text {max }}=1689(\mathrm{CO}) \mathrm{cm}^{-1}$; MS (EI, 70 EV ): $\mathrm{m} / \mathrm{z}$ $(\%)=237\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta_{\mathrm{H}}=6.11$ (s, 2 H , $\mathrm{CH}_{2}$ ), 7.37-7.41 (m, 2H, arom. H), 7.47-7.57 (m, 3H, arom. H), 7.66-7.70 (m, 1H, arom. H), 8.06-8.11 (m, 3H, arom. H).

Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}$ (237.25): C, 70.87; H, 4.67; N, 17.71. Found C, 70.90; H, 4.53; N, 17.69.

General Procedure for the Preparation of Compounds 5a-f.
To a stirred suspension of each of compound 3a-c ( 10 mmol ) in ethanol ( 20 ml ) containing potassium hydroxide $(0.56 \mathrm{~g}, 10$ $\mathrm{mmol})$, each of compound $\mathbf{1 a}, \mathbf{b}(10 \mathrm{mmol})$ was added. The reaction mixture was heated under reflux for 2 h . The solvent was evaporated under reduced pressure and the remaining product was triturated with water and neutralized with hydrochloric acid to deposit a solid, which was crystallized from ethanol/dioxane (3:1).

[^0]This compound was obtained in $77 \%$ yield ( 3.01 g ); mp. $217^{\circ}$; ir $(\mathrm{KBr}) v_{\text {max }}=3432(\mathrm{OH}) \mathrm{cm}^{-1}$; MS (EI, 70 EV$): \mathrm{m} / \mathrm{z}(\%)=391$ $\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta_{\mathrm{H}}=7.37\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{2}\right), 7.48-$ $7.56(\mathrm{~m}, 7 \mathrm{H}$, arom. H), 7.59-7.64 (m, 4H, arom. H), 7.67-7.69 (m, 2 H , arom. H), $7.99\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{5}\right), 8.19$ (d, $1 \mathrm{H}, J=8.4 \mathrm{~Hz}$, benzo-triazolyl-H), 11.55 (s, 1H, OH); ${ }^{13} \mathrm{C}$ NMR (deuteriochloroform): $\delta_{\mathrm{C}}=156.01,152.99,145.16,144.07,143.90,142.46,138.44$, $135.00,131.76,131.37,130.24,128.79,124.20,124.04,123.39$, 122.97, 120.51, 118.92, 115.64, 111.85.

Anal. Calcd. for $\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}$ (391.42): C, 73.64; H, 4.38; N , 17.89. Found C, 73.64; H, 4.53; N, 17.62.

2-Benzotriazol-1-yl-4-phenylazo-5-thiophen-2-yl-phenol (5b).
This compound was obtained in $92 \%$ yield ( 3.65 g ); mp. $280^{\circ}$; ir (KBr) $v_{\max }=3422(\mathrm{OH}) \mathrm{cm}^{-1}$; MS (EI, 70 EV ): $m / z$ $(\%)=397\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=7.28(\mathrm{dd}, 1 \mathrm{H}, J=4 \mathrm{~Hz}$, thienyl $\mathrm{H}_{4}$ ), 7.47-7.51 (m, 1 H , thienyl $\mathrm{H}_{3}$ ), $7.56(\mathrm{~d}, 1 \mathrm{H}, J=8.0$ Hz , benzotriazolyl-H), 7.59-7.65 (m, 5H, Ph), $7.68\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{2}\right)$, $7.86\left(\mathrm{dd}, 1 \mathrm{H}, J=4.6 \mathrm{~Hz}\right.$, thienyl $\left.\mathrm{H}_{5}\right), 7.93(\mathrm{~m}, 2 \mathrm{H}$, benzotria-zolyl-H), $7.97\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{5}\right), 8.18(\mathrm{~d}, 1 \mathrm{H}, J=8.2 \mathrm{~Hz}$, benzotriazolyl H), $11.68(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{C}}=155.68$, 153.01, 145.91, 141.17, 138.19, 137.15, 134.37, 132.28, 132.17, 130.42, 129.22, 128.98, 128.35, 125.17, 124.43, 124.11, 120.22, 116.62, 116.32, 112.63.

Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{OS}$ (397.38): C, $66.49 ; \mathrm{H}, 3.80 ; \mathrm{N}$, 17.63; S, 8.05. Found C, 66.72; H, 4.08; N, 17.52; S, 7.84.

## 2-Benzotriazol-1-yl-5-furan-2-yl-4-phenylazo-phenol (5c).

This compound was obtained in $91 \%$ yield ( 3.46 g ); mp. $260^{\circ}$; ir (KBr) $v_{\text {max }}=3435(\mathrm{OH}) \mathrm{cm}^{-1}$; MS (EI, 70 EV ): $m / z$ $(\%)=381\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=6.76-6.78(\mathrm{~m}, 1 \mathrm{H}$, furyl $\mathrm{H}_{4}$ ), $7.08\left(\mathrm{~d}, 1 \mathrm{H}, J=3.4 \mathrm{~Hz}\right.$, furyl $\left.\mathrm{H}_{3}\right), 7.47-7.51(\mathrm{~m}, 1 \mathrm{H}$, furyl $\mathrm{H}_{5}$ ), 7.56-7.64 (m, 5H, Ph), $7.76\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{6}\right), 7.89(\mathrm{~m}, 2 \mathrm{H}$, benzotriazolyl H), $7.93\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{3}\right), 8.01-8.02(\mathrm{~m}, 1 \mathrm{H}$, benzo-triazolyl-H), $8.17(\mathrm{~d}, 1 \mathrm{H}, J=8.1 \mathrm{~Hz}$, benzotriazolyl H$), 11.54$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{OH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (DMSO): $\delta_{\mathrm{C}}=155.56,153.08,149.58$, $145.89,145.22,140.97,134.35,132.57,132.18,130.49$, $128.95,125.16,123.94,123.71,120.21,116.44,116.38$, 114.02, 113.97, 112.65.

Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{2}$ (381.38): C, 69.28; H, 3.96; N , 18.36. Found C, 69.18; H, 4.11; N, 18.13.

## 4-Benzimidazol-1-yl-6-phenylazo-biphenyl-3-ol (5d).

This compound was obtained in $80 \%$ yield ( 3.12 g ); mp. $288^{\circ}$; ir ( KBr ) $v_{\text {max }}=3420 \mathrm{~cm}^{-1}(\mathrm{OH})$; MS (EI, 70 EV$): \mathrm{m} / \mathrm{z}(\%)=390$ $\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \operatorname{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=7.27\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{2}\right), 7.29-7.30(\mathrm{~m}, 2 \mathrm{H}$, arom. H), 7.38-7.40 (m, 1H, arom. H), 7.46-7.57 (m, 8H, arom. $\mathrm{H}), 7.64-7.66(\mathrm{~m}, 2 \mathrm{H}$, arom. H), 7.77-7.79 (m, 1 H , arom. H), 7.90 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H}_{5}$ ), $8.47\left(\mathrm{~s}, 1 \mathrm{H}\right.$, benzimidazolyl $\left.\mathrm{H}_{2}\right), 11.20(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH})$.

Anal. Calcd. for $\mathrm{C}_{25} \mathrm{H}_{18} \mathrm{~N}_{4} \mathrm{O}$ (390.43): C, 76.90; H, 4.65; N, 14.35. Found C, $77.09 ;$ H, $4.73 ; \mathrm{N}, 14.26$.

2- Benzimidazol-1-yl-4-phenylazo-5-thiophen-2-yl-phenol (5e).
This compound was obtained in $85 \%$ yield ( 3.36 g ); mp. $276^{\circ}$; ir $(\mathrm{KBr}) \mathrm{v}_{\text {max }}=3439(\mathrm{OH}) \mathrm{cm}^{-1}$; MS (EI, 70 EV$): \mathrm{m} / \mathrm{z}(\%)=396$ $\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=7.25-7.31(\mathrm{~m}, 3 \mathrm{H}$, arom. H$), 7.37-$ $7.40\left(\mathrm{~m}, 1 \mathrm{H}\right.$, thienyl $\left.\mathrm{H}_{4}\right), 7.54\left(\mathrm{~d}, 1 \mathrm{H}, J=4.8 \mathrm{~Hz}\right.$, thienyl $\left.\mathrm{H}_{3}\right), 7.62-$ $7.35(\mathrm{~m}, 4 \mathrm{H}$, arom. H), 7.76-7.78 (m, 1H, arom. H), $7.84(\mathrm{~d}, 1 \mathrm{H}, J=$ 4.6 Hz , thienyl $\mathrm{H}_{5}$ ), $7.88(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-3), 7.92(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}$, benzimidazolyl H ), $8.46\left(\mathrm{~s}, 1 \mathrm{H}\right.$, benzimidazolyl $\mathrm{H}_{2}$ ), $11.40(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH})$; ${ }^{13} \mathrm{C} \mathrm{nmr}$ (DMSO): $\delta_{\mathrm{C}}=156.46,153.03,145.10,143.82,141.11$,
$138.40,136.16,134.91,132.02,130.40,128.89,128.28,124.26$, $124.09,123.03,121.96,120.49,118.48,116.49,115.76,111.87$.

Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{OS}$ (396.38): C, 69.69; H, 4.07; N, 14.14; S, 8.07. Found C, 69.63; H, 4.18; N, 13.95; S, 7.83.

## 2- Benzimidazol-1-yl-5-furan-2-yl-4-phenylazo-phenol (5f).

This compound was obtained in $68 \%$ yield ( 2.58 g ); mp. $273^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3435(\mathrm{OH}) \mathrm{cm}^{-1}$; MS $(\mathrm{EI}, 70 \mathrm{EV}): m / z(\%)=380$ $\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=6.75\left(\mathrm{t}, 1 \mathrm{H}, J=3.4 \mathrm{~Hz}\right.$, furyl $\left.\mathrm{H}_{4}\right)$, $7.03\left(\mathrm{~d}, 1 \mathrm{H}, J=3.4 \mathrm{~Hz}\right.$, furyl $\left.\mathrm{H}_{3}\right), 7.28-7.30(\mathrm{~m}, 2 \mathrm{H}$, arom. H), 7.36-7.38 (m, 1H, furyl $\left.\mathrm{H}_{5}\right), 7.55-7.63(\mathrm{~m}, 3 \mathrm{H}$, arom. H$), 7.71(\mathrm{~s}$, 1H, H-6), 7.76-7.82 (m, 1H, benzoimidazolyl H), 7.84-7.87 (m, 3 H , benzoimidazolyl H), $7.98(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-3), 8.46(\mathrm{~s}, 1 \mathrm{H}$, benzoimidazolyl $\mathrm{H}_{2}$ ), $11.55(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH})$.

Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}_{2}$ (380.39): C, $72.62 ; \mathrm{H}, 4.24 ; \mathrm{N}$, 14.73. Found C, 72.88 ; H, 4.25; N, 14.56.

General Procedure for the Preparation of Compounds 8a-c.
Similar reaction conditions as for compounds 7a-f using compound $2(10 \mathrm{mmol})$ were utilized. The target product $\mathbf{8 a - c}$ was crystallized from ethanol/dioxane (3:1).

## 6-Phenylazo-4-[1,2,4]-triazol-1-yl-biphenyl-3-ol (8a).

This compound was obtained in $88 \%$ yield ( 3.0 g ); mp. $230^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3415(\mathrm{OH}) \mathrm{cm}^{-1} ;$ MS $(\mathrm{EI}, 70 \mathrm{EV}): m / z(\%)=341$ $\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta_{\mathrm{H}}=7.38\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{2}\right), 7.46-$ $7.51(\mathrm{~m}, 6 \mathrm{H}$, arom. H), 7.55-7.57 (m, 2H, arom. H), 7.79-7.81 (m, 2 H , arom. H$), 8.11\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{5}\right), 8.27\left(\mathrm{~s}, 1 \mathrm{H}\right.$, triazolyl $\left.\mathrm{H}_{4}\right), 8.89(\mathrm{~s}$, 1 H , triazolyl $\mathrm{H}_{2}$ ), $10.61(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH})$.
Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$ (341.36): C, 70.37; H, 4.43; N , 20.52. Found C, $70.31 ;$ H, $4.58 ; \mathrm{N}, 9.87$.

4-Phenylazo-5-thiophen-2-yl-2-[1,2,4]-triazol-1-yl-phenol (8b).
This compound was obtained in $85 \%$ yield ( 3.0 g ); mp. $271^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3435(\mathrm{OH}) \mathrm{cm}^{-1} ; \mathrm{MS}(\mathrm{EI}, 70 \mathrm{EV}): \mathrm{m} / \mathrm{z}(\%)=396$ $\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=7.23\left(\mathrm{t}, 1 \mathrm{H}, J=4 \mathrm{~Hz}\right.$, thienyl $\left.\mathrm{H}_{4}\right)$, 7.54-7.64 (m, 5H, arom. H), $7.82\left(\mathrm{~d}, 1 \mathrm{H}, J=4.2 \mathrm{~Hz}\right.$, thienyl $\mathrm{H}_{5}$ ), 7.94-7.96 (m, 2 H , arom. H), $8.15\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{3}\right), 8.32(\mathrm{~s}, 1 \mathrm{H}$, triazolyl $\mathrm{H}_{4}$ ), $9.15\left(\mathrm{~s}, 1 \mathrm{H}\right.$, triazolyl $\left.\mathrm{H}_{2}\right), 11.71(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH})$.

Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{OS}$ (347.32): C, 62.24; H, 3.78; N, 20.17; S, 9.21. Found C, 62.48; H, 3.99; N, 19.95; S, 9.11.

5-Furan-2-yl- 4-phenylazo-2-[1,2,4]-triazol-1-yl-phenol (8d).
This compound was obtained in $87 \%$ yield ( 2.85 g ); mp. $236^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3435(\mathrm{OH}) \mathrm{cm}^{-1} ; \mathrm{MS}(\mathrm{EI}, 70 \mathrm{EV}): m / z(\%)=380$ $\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=6.79\left(\mathrm{t}, 1 \mathrm{H}, J=3.6 \mathrm{~Hz}\right.$, furyl $\left.\mathrm{H}_{4}\right)$, $7.00\left(\mathrm{~d}, 1 \mathrm{H}, J=3.6 \mathrm{~Hz}\right.$, furyl $\left.\mathrm{H}_{3}\right), 7.54-7.65(\mathrm{~m}, 4 \mathrm{H}$, arom. H), 7.81-7.96 (m, 3H, arom. H), $8.12\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{3}\right), 8.26(\mathrm{~s}, 1 \mathrm{H}$, triazolyl H-4), 9.16 (s, 1H, triazolyl $\mathrm{H}_{2}$ ), $11.70(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH})$.

Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}_{2}$ (331.32): C, 65.25; H, 3.96; N, 21.14. Found C, $65.09 ; \mathrm{H}, 4.07 ; \mathrm{N}, 21.09$.

General Procedure for the Preparation of Compounds $\mathbf{1 1 a , b}$ and 13a,b.

A stirred suspension of each of compound $\mathbf{1 b}$ and $2(10 \mathrm{mmol})$ and phenyl isothiocyanate $(1.35 \mathrm{~g}, 10 \mathrm{mmol})$ in DMF ( 20 ml ) in the presence of potassium hydroxide $(0.06 \mathrm{~g})$; $\alpha$-haloketone ( 10 mmol ) is added to the stirred solution after 10 h and the reaction mixture is heated under reflux for 4 h . The solvent was removed and the residue was triturated with water and neutralized with hydrochloric acid. The solid product was crystallized from ethanol.

1-(4-Benzimidazol-1-yl-3-methyl-5-phenylaminothiophen-2yl)ethanone 11a.

This compound was obtained in $75 \%$ yield ( 2.60 g ); mp. $240^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3086(\mathrm{NH}), 1614(\mathrm{CO}) \mathrm{cm}^{-1} ; \mathrm{MS}(\mathrm{EI}, 70 \mathrm{EV})$ : $m / z(\%)=347\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform) $\delta_{\mathrm{H}}=2.10(\mathrm{~s}$, $\left.3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.52\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{COCH}_{3}\right), 7.10-7.14(\mathrm{~m}, 1 \mathrm{H}$, arom. H), 7.17-7.23 (m, 2H, arom. H), 7.36-7.41 (m, 3H, arom. H), 7.46$7.51(\mathrm{~m}, 3 \mathrm{H}$, arom. H$), 8.96(\mathrm{~s}, 1 \mathrm{H}$, benzimidazolyl-H), $9.57(\mathrm{br}$ $\mathrm{s}, 1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deutereochloroform): $\delta_{\mathrm{C}}=189.9(\mathrm{CO})$, $151.8,143.6,143.0,142.4,141.3,134.0,130.0,124.5,124.1$, 123.7, 120.2, 119.8, 119.2, 115.7, 110.7, 29.5, 14.5.

Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{OS}$ (347.36): C, $69.15 ; \mathrm{H}, 4.93$; N, 12.10; S, 9.21. Found C, 68.95; H, 5.01; N, 12.08; S, 9.00.
(4-Benzimidazol-1-yl-3-methyl-5-phenylaminothiophen-2-yl)phenylmethanone (11b).

This compound was obtained in $69 \%$ yield ( 2.82 g ); mp. $150^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3055(\mathrm{NH}), 1606(\mathrm{CO}) \mathrm{cm}^{-1} ; \mathrm{MS}(\mathrm{EI}, 70 \mathrm{EV})$ : $m / z(\%)=409\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=1.83\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$, 7.01-7.03 (m, 1H, arom. H), 7.26-7.32 (m, 7H, arom. H), 7.50$7.57(\mathrm{~m}, 3 \mathrm{H}$, arom. H), 7.69-7.71 (m, 2H, arom. H), 7.76-7.78 (m, 1 H , arom. H ), $8.38\left(\mathrm{~s}, 1 \mathrm{H}\right.$, benzimidazolyl $\left.\mathrm{H}_{2}\right), 9.36(\mathrm{br} \mathrm{s}, 1 \mathrm{H}$, $\mathrm{NH}) ;{ }^{13} \mathrm{C} \operatorname{nmr}(\mathrm{DMSO}): \delta_{\mathrm{C}}=188.3(\mathrm{CO}), 153.3,145.6,144.6$, $142.5,141.8,140.8,134.6,132.5,130.31,129.4,128.8,124.7$, $124.3,123.6,120.1,119.8,118.4,117.4,111.5,15.0$.

Anal. Calcd. for $\mathrm{C}_{25} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{OS}$ (409.43): C, $73.33 ; \mathrm{H}, 4.68 ; \mathrm{N}$, 10.26; S, 7.81. Found C, 73.11; H, 4.74; N, 10.39; S, 7.60.

1-(3-Methyl-5-phenylamino-4-[1,2,4]-triazol-1-yl-thiophen-2yl)ethanone (13a).

This compound was obtained in $80 \%$ yield $(2.40 \mathrm{~g})$; mp. $185^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3064(\mathrm{NH}), 1613(\mathrm{CO}) \mathrm{cm}^{-1} ; \mathrm{MS}(\mathrm{EI}, 70 \mathrm{EV})$ : $m / z(\%)=298\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta_{\mathrm{H}}=2.05(\mathrm{~s}$, $\left.3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.43\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{COCH}_{3}\right), 7.04-7.07(\mathrm{~m}, 1 \mathrm{H}$, arom. H), 7.27-7.36 (m, 5H, arom. H), $8.26\left(\mathrm{~s}, 1 \mathrm{H}\right.$, triazolyl $\left.\mathrm{H}_{4}\right), 8.77$ (s, 1 H , triazolyl $\mathrm{H}_{2}$ ), 9.16 (br s, $1 \mathrm{H}, \mathrm{NH}$ ).

Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{OS}$ (298.29): C, 60.39; H, 4.73; N , 18.78; S, 10.73. Found C, 60.59; H, 4.79; N, 18.80; S, 10.71.
(3-Methyl-5-phenylamino-4-[1,2,4]-triazol-1-yl- -thiophen-2-yl)phenyl-methanone (13b).

This compound was obtained in $72 \%$ yield ( 2.60 g ); mp. $190^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3069(\mathrm{NH}), 1617(\mathrm{CO}) \mathrm{cm}^{-1}$; MS (EI, 70 EV$)$ : $m / z(\%)=360.36\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} n m r(\mathrm{DMSO}): \delta_{\mathrm{H}}=1.88(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{CH}_{3}\right), 7.05-7.07(\mathrm{~m}, 1 \mathrm{H}$, arom. H), 7.26-7.35 (m, 7 H , arom. H), 7.49-7.53 (m, 3 H , arom. H), 7.57-7.59 (m, 1H, arom. H), 7.65$7.67(\mathrm{~m}, 2 \mathrm{H}$, arom. H), $8.26(\mathrm{~s}, 1 \mathrm{H}$, triazolyl H 4$), 8.80(\mathrm{~s}, 1 \mathrm{H}$, triazolyl $\mathrm{H}_{2}$ ), 9.29 (br s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{C}}=188.47$, 153.33, 153.28, 147.62, 144.20, 141.51, 140.71, 132.54, 130.45, $129.44,128.75,124.66,120.12,119.31,117.99,14.87\left(\mathrm{CH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{OS}$ (360.36): C, 66.66; H, 4.48; N, 15.55; S, 8.88. Found C, 67.00; H, 4.48; N, 15.40; S, 9.06.

5-Benzotriazol-1-yl-2-oxo-4,6-diphenyl-1,2-dihydropyridine-3carbonitrile (17).

Benzylidenemalononitrile $(1.54 \mathrm{~g}, 10 \mathrm{mmol})$ was added to a stirred suspension of compound $\mathbf{1 c}(2.37 \mathrm{~g}, 10 \mathrm{mmol})$ in ethanol $(20 \mathrm{ml})$ in the presence of sodium hydroxide $(0.5 \mathrm{~g})$. The reaction mixture was heated under reflux for 2 h . The solvent was reduced under vacuum; poured onto water and the residue was neutralized
with diluted hydrochloric acid solution to deposit a solid, which was crystallized from dioxane. Yield: $2.40 \mathrm{~g}(62 \%) ; \mathrm{mp} .>300^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=3060(\mathrm{NH}), 2229(\mathrm{CN}), 1672(\mathrm{CO}) \mathrm{cm}^{-1}$; MS $(\mathrm{EI}, 70 \mathrm{EV}): m / z(\%)=389\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=7.00-$ $7.26(\mathrm{~m}, 10 \mathrm{H}$, arom. H), 7.33-7.40 (m, 2H, benzotriazolyl-H), $7.60(\mathrm{~d}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}$, benzotriazolyl-H), $7.78(\mathrm{~d}, 1 \mathrm{H}, J=8.6$ Hz , benzotriazolyl-H), 13.46 (br s, 1H, NH).

Anal. Calcd. for $\mathrm{C}_{24} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$ (389.40): C, $74.02 ; \mathrm{H}, 3.88 ; \mathrm{N}$, 17.99. Found C, 74.27; H, 4.02; N, 18.21.

5-Benzotriazol-1-yl-6-methyl-2-thioxo-1,2-dihydropyridine-3carbonitrile (20).

Cyanothioacetamide ( $1 \mathrm{~g}, 10 \mathrm{mmol}$ ) was added to a stirred suspension of the enaminone $18(2.30 \mathrm{~g}, 10 \mathrm{mmol})$ in ethanol ( 20 ml ) in the presence of few drops of triethyl amine. The reaction mixture was heated under reflux for 2 h . The solvent was reduced under vacuum; poured onto water and the residue was neutralized with diluted hydrochloric acid solution to deposit a solid, which was crystallized from ethanol. Yield: $1.70 \mathrm{~g}(64 \%)$; mp. $290^{\circ}$; ir ( KBr ) $v_{\text {max }}=3013(\mathrm{NH}), 2224(\mathrm{CN}) \mathrm{cm}^{-1} ;$ MS (EI, 70 EV$): m / z(\%)=267$ $\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} n m r$ (deuteriochloroform): $\delta=2.38\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 7.36-$ $7.38(\mathrm{~m}, 1 \mathrm{H}$, benzotriazolyl-H), 7.52-7.59 $(\mathrm{m}, 1 \mathrm{H}$, benzotriazolyl$\mathrm{H}), 7.62-7.69\left(\mathrm{~m}, 1 \mathrm{H}\right.$, benzotriazolyl-H), $8.01\left(\mathrm{~s}, 1 \mathrm{H}\right.$, pyridyl $\left.\mathrm{H}_{4}\right)$, 8.18-8.27 (m, 1H, benzotriazolyl-H), 13.65 (br s, 1H, NH).

Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{~S}$ (267.24): C, $58.42 ; \mathrm{H}, 3.39$; N, $26.21 ;$ S, 11.97. Found: C, 58.51; H, 3.54; N, 25.95; S, 11.82.

5-Benzotriazol-1-yl-6-methyl-2-(2-oxopropylsulfanyl)nicotinonitrile (21).

Chloroacetone ( $0.92 \mathrm{~g}, 10 \mathrm{mmol}$ ) was added to a stirred suspension of compound $20(2.67 \mathrm{~g}, 10 \mathrm{mmol})$ in DMF $(20 \mathrm{ml})$ in the presence of potassium hydroxide $(0.5 \mathrm{~g})$. The reaction mixture was heated under reflux for 6 h . The solvent was reduced under vacuum; poured onto water and the residue was neutralized with diluted hydrochloric acid solution to deposit a solid, which was crystallized from ethanol/dioxane (1:2). Yield: 2.60 g (80 $\%) ; \mathrm{mp} .259^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=2228(\mathrm{CN}), 1676(\mathrm{CO}) \mathrm{cm}^{-1}$; MS (EI, 70 EV ): $m / z(\%)=323\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta_{\mathrm{H}}=2.46\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{3}\right), 2.48\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{COCH}_{3}\right), 6.79\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, $7.34(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}$, benzotriazolyl-H), $7.50(\mathrm{t}, 1 \mathrm{H}, J=8.6$ Hz , benzotriazolyl-H), $7.59(\mathrm{t}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}$, benzotriazolyl-H), $8.08\left(\mathrm{~s}, 1 \mathrm{H}\right.$, pyridyl $\left.\mathrm{H}_{4}\right), 8.20(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}$, benzotriazolyl$\mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{nmr}(\mathrm{DMSO}): \delta_{\mathrm{C}}=192.37(\mathrm{CO}), 160.68,157.78$, $148.15,145.86,134.42,131.60,129.82,128.79,125.73,120.58$, 111.62, 106.04, $31.57\left(\mathrm{CH}_{2}\right), 29.89\left(\mathrm{CH}_{3}\right), 22.11\left(\mathrm{COCH}_{3}\right)$.

Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{OS}$ (323.30): C, $59.44 ; \mathrm{H}, 4.05$; N, 21.66; S, 9.90. Found C, 59.64; H, 4.07; N, 21.66; S, 9.68.
$N^{\prime}$ '(2-Acetyl-5-benzotriazol-1-yl-6-methyl-thieno[2,3- $b$ ]pyridin-$3-\mathrm{yl})-\mathrm{N}, \mathrm{N}$-dimethylformamidine (22).

To a suspension of compound $20(3.23 \mathrm{~g}, 10 \mathrm{mmol})$ in xylene (20 $\mathrm{ml})$, dimethylformamide dimethylacetal $(1,19 \mathrm{~g}, 10 \mathrm{mmol})$ was added. The reaction mixture was heated under reflux for 4 h . The solvent was removed and the residue cooled to deposit a solid, which was crystallized from dioxane/ethanol (1:1). Yield: 2.50 g (65 \%); mp. $264^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=1636(\mathrm{CO}) \mathrm{cm}^{-1}$; MS (EI, 70 EV$): m / z$ $(\%)=378\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=2.29\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.55(\mathrm{~s}$,
$\left.3 \mathrm{H}, \mathrm{COCH}_{3}\right), 3.01\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{NMe}_{2}\right), 7.53(\mathrm{t}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}$, benzo-triazolyl-H), 7.61-7.67 (m, 2H, benzotriazolyl-H), $7.97\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{4}\right)$, $8.23(\mathrm{~d}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}$, benzotriazolyl-H), $8.42(\mathrm{~s}, 1 \mathrm{H}$, amidine- H$)$.

Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{18} \mathrm{~N}_{6} \mathrm{OS}$ (378.38): C, 60.31; H, 4.80; N , 22.21 ; S, 8.46. Found C, 60.14; H, 4.88; N, 21.98; S, 8.21.

8-Benzotriazol-1-yl-7-methyl-1 $H$-thieno[2,3- $b$; 4,5-b']dipyridin-4-one (23).

To a suspension of compound $22(3.78 \mathrm{~g}, 10 \mathrm{mmol})$ in dioxane $(20 \mathrm{ml})$, sodium hydride $(0.4 \mathrm{~g}, 60 \%$ dispersion $)$ was added. The reaction mixture was heated under reflux for 6 h . The solvent was reduced under vacuum; poured into water and the residue was neutralized with diluted hydrochloric acid solution to deposit a solid, which was crystallized from dioxane/ethanol (1:3). Yield: $2.0 \mathrm{~g}(60 \%) ; \mathrm{mp} .>300^{\circ}$; ir $(\mathrm{KBr}) v_{\max }=33.92(\mathrm{OH}) \mathrm{cm}^{-1} ; \mathrm{MS}$ $(\mathrm{EI}, 70 \mathrm{EV}): m / z(\%)=333\left[\mathrm{M}^{+}\right] ;{ }^{1} \mathrm{H} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{H}}=2.48(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $3.84(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}), 6.62\left(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}, \mathrm{H}_{3}\right), 7.57$ $(\mathrm{t}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}$, benzotriazolyl-H), $7.68(\mathrm{t}, 1 \mathrm{H}, J=8.6 \mathrm{~Hz}$, benzotriazolyl-H), $7.76\left(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}, \mathrm{H}_{2}\right), 8.26-8.30(\mathrm{~m}$, 2 H , benzotriazolyl-H), $9.04\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{9}\right) ;{ }^{13} \mathrm{C} \mathrm{nmr}(\mathrm{DMSO}): \delta_{\mathrm{C}}=$ 161.23 (CO), 156.97, 145.92, 142.65, 134.41, 130.62, 129.93, $129.50,128.45,127.65,126.21,125.82,124.79,120.65,111.97$, 111.59, 22.23.

Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{11} \mathrm{~N} 5 \mathrm{OS}$ (333.38): C, 61.26; H, 3.33; N, 21.01; S, 9.60. Found C, 61.14; H, 3.48; N, 20.92; S, 9.87.

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[^0]:    4-Benzotriazol-1-yl-6-phenylazobiphenyl-3-ol (5a).

